Synthesis of Isoindole by Retro-Diels-Alder Reaction

By J. BORNSTEIN* and D. E. REMY

(Department of Chemistry, Boston College, Chestnut Hill, Massachusetts 02167)

and JOAN E. SHIELDS

(Department of Chemistry, C. W. Post College, Brookville, New York 11548)

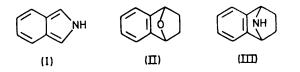
Summary Thermal fragmentation of 1,2,3,4-tetrahydro-1,4-epiminonaphthalene at reduced pressure affords pure isoindole in essentially quantitative yield.

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THE first isolation of isoindole (I) was reported recently by Bonnett and Brown,¹ who prepared it by pyrolysis of 2-(methoxycarbonyloxy)isoindoline at 500° and 0.01 mmHg. At about the same time, Wiersum and Mijs described² a simple route to isobenzofuran involving the thermal fragmentation at reduced pressure of the epoxynaphthalene (II); ethylene is formed as coproduct in this 4 + 2 retrocyclo-addition. Since the preparative work of Bonnett and Brown demonstrated that isoindole can tolerate short exposure to elevated temperature *in vacuo*, it seemed reasonable that this elusive molecule could also be synthesized by vacuum thermolysis of the epiminonaphthalene (III)³, the nitrogen analogue of (II). We now report that this is, indeed, the case.

Sublimation of (III) (ca. 1 g per 20 min) from a glass tube at 80° and 0.5 mmHg through an unpacked quartz tube

(300 mm; 24 mm diam.) at 600° afforded isoindole (I) and ethylene. The products were collected in a trap cooled by liquid nitrogen. Subsequent replacement of the nitrogen coolant by a solid CO₂-acetone bath caused the ethylene to distil leaving only the white crystals of isoindole



(I) in essentially quantitative yield. Ethylene was characterized by passing it into a solution of bromine in CCl₄. The resulting ethylene dibromide was shown by i.r. and n.m.r. spectroscopy to be identical with an authentic specimen. Evidence that the isoindole (I) was pure and had, therefore, successfully survived the conditions of its

- ¹ R. Bonnett and R. F. C. Brown, J.C.S. Chem. Comm., 1972, 393.
- ² U. E. Wiersum and W. J. Mijs, *J.C.S. Chem. Comm.*, 1972, 347.
 ³ L. A. Carpino and D. E. Barr, *J. Org. Chem.*, 1966, 31, 764.
 ⁴ R. Kreher and J. Seubert, *Z. Naturforsch.*, 1965, 20b, 75.

production was obtained by both spectral and chemical means. The n.m.r., i.r., u.v., and mass spectra of (I) were identical with those reported by Bonnett and Brown.¹ As expected, (I) gives a colour (magenta) with Ehrlich's reagent and decomposes rapidly at room temperature.

Treatment of cold crystals of (I) with ethereal N-phenylmaleimide afforded, after recrystallization from ethanol, the exo-adduct, m.p. 208-209° (70%), shown by t.l.c., m.p., mixed m.p., and n.m.r. analysis to be identical with a specimen prepared by the method of Kreher and Seubert.⁴

This preparation of isoindole appears to have some advantage over that employing 2-(methoxycarbonyloxy)isoindoline since it affords (I) free of coproducts.

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